Synthesis of Fused Purinoquinazolines. Three New Heterocyclic Ring Systems.

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The Ullmann reaction of 8-aminotheophylline or 8-aminocaffeine with 2-chlorobenzoic acid and of 8-bromotheophylline with ethyl-2-aminobenzoate afforded derivatives of three new heterocyclic systems: purino[7,8-a]quinazoline-5,9,11(6H,8H,10H)-trione, purino[8,9-b]quinazoline-2,4,11(1H,3H,5H)-trione and purino[8,7-b]quinazoline-2,4,6(1H,3H,11H)-trione, respectively.

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Pursuing our interest in the field of new heteropoly-cyclic compounds which can be related to important alkaloids containing the indole nucleus as a part of a poly-cyclic system, in the past we have described the synthesis of some 6H-indolo[2,3-b][1,8]naphthyridines 1 [1] and, more recently, of some 5H,12H-[1]benzoxepino[4,3-b]-indol-6-ones 2 [2].

In the last few years we have also reported the synthesis of molecules which contain new heterocyclic ring systems, such as a number of 5,7-dihydro-5-oxopyrido[3',2':5,6]pyrimido[1,2-a]benzimidazoles 3 [3] and 11-alkyl-5,11-dihydro-5-oxopyrido[2',3':4,5]pyrimido[1,2-a]benzimidazoles 4 [4].

As part of our studies on new planar heteropolycyclic compounds as new potential antitumor agents, we wish now to report the synthesis of some purinoquinazoline derivatives: 8,10-dimethylpurino[7,8-a]quinazoline-5,9,11(6H,8H,10H)-triones 6, 8 and 9, 1,3-dimethylpurino[8,7-b]quinazoline-2,4,6(1H,3H,11H)-triones 11, 13 and 14, and 1,3,5-trimethylpurino[8,9-b]quinazoline-2,4,11(1H,3H,5H)-trione 17, each one of these structures representing a new heterocyclic ring system.

The target compound 6 was synthesized in 40% yield as shown in Scheme 1, *via* an Ullmann reaction between 2-chlorobenzoic acid and 8-aminotheophylline 5 [5,6], in

the presence of anhydrous potassium carbonate and a catalytic amount of cuprous bromide. The structure of 6 was confirmed by analytical, ir, ¹H nmr and mass (M+, m/z = 297) spectral data (Table 1) and chemical evidence. When 8-aminotheophylline 5 and an equimolar amount of 2chlorobenzoic acid were heated at 120° with an excess of polyphosphoric acid (PPA), the benzamide 7 was obtained in 45% yield (Scheme 1). Its structure was confirmed by analytical, ir and ¹H nmr spectral data. The cyclization reaction of 7 to 6 was effected by refluxing 7 for 4 hours in DMF in the presence of anhydrous potassium carbonate [7] (Scheme 1). Crude 6 can easily be purified by sublimation and recrystallization from DMF (51% yield). The preparation of 6 from 7 confirms the structure proposed for 6, because the cyclization of the amide 7 on the N(9) of theophylline is not possible, due to the steric hindrance of the 3-methyl group (Dreiding models) [8].

Compound 6 was methylated to 8 with dimethyl sulphate in acetonitrile solution, in the presence of potassium carbonate [9], in a satisfactory yield (68%) (Scheme 1). The structure of 8 was confirmed by analytical, ir, ¹H nmr and mass data (Table 1). On the other hand, methylation at the 7 position is not possible because of the steric hindrance of the 8-methyl group [8].

When 6 was refluxed for 10 hours in a small volume of DMSO, a mixture of 6 and of the methylthiomethyl derivative 9 was obtained (Scheme 1), which was separated by preparative tlc. A similar reaction is reported in literature for amide compounds such as isatin, phthalimide, saccharin, etc. [10-12]. The structure of compound 9 was mainly confirmed by elemental analysis, which revealed the presence of sulfur in the molecular formula, by the 1 H nmr spectrum which showed, in addition to the singlets relative to the methyl groups at the positions 8 and 10, also another CH₃ at $\delta = 2.35$ ppm and a CH₂ at $\delta = 5.42$ ppm, and by the mass spectrum in which the base peak at m/z = 61 was due to the CH₂SCH₃ fragment (Table 1).

The 1,3-dimethylpurino[8,7-b]quinazoline-2,4,6-(1H,3H,11H)-trione 11, an isomer of 6, was prepared in 45% yield *via* an Ullmann reaction between 8-bromotheophylline

7

Scheme 2

Scheme 3

18 19

Table 1
Physical and Spectral Data of Compounds 6, 8 and 9

6, 8, 9

Compoun No.	nd R	Yield (%)	Mp (°C) (recrystallization solvent)	IR (cm ⁻¹)	¹ H-NMR (δ ppm) (DMSO-d ₆) [a]	MS m/z (R.I. %)	Molecular Formula	Analysis(%) Calcd./Found		
								С	Н	N
6	Н	40	>300 (DMF)	3150, 1680, 1640, 1600,	3.32 (s, 3H, 10-CH ₃), 3.47 (s, 3H, 8-CH ₃),	M+297 (100)	$C_{14}H_{11}N_5O_3$	56.57 56.48	3.73 3.81	23.56 23.65
			•	1510, 1220, 1130, 760.	7.45-9.65 (m, 4H, Ar-H).					
8	CH_3	68	289-290	1670, 1640,	3.34 (s, 3H, 10-CH ₃),	M+311	$C_{15}H_{13}N_5O_3$	57.88	4.21	22.50
			(DMF)	1590, 1510, 1430, 1230, 980, 760, 750.	3.52 (s, 3H, 8-CH ₃), 3.67 (s, 3H, 6-CH ₃), 7.50-9.65 (m, 4H, Ar-H).	(100)		58.00	4.25	22.38
9	CH ₂ SCH ₃	56	278-280 (DMSO)	1690, 1640,	2.35 (s, 3H, S-CH ₃), 3.37	M+357	$C_{16}H_{15}N_5O_3S$	53.77	4.23	19.60
				1600, 1580,	(s, 3H, 10-CH ₃), 3.54 (s,	(7),		53.61	4.19	19.69
				1540, 1500,	3H, 8-CH ₃), 5.42 (s, 2H,	61				
				1220, 980, 760, 740.	CH ₂ -S), 7.55-9.75 (m, 4H, Ar-H).	(100)				

[a] Recorded on a Bruker AC-200.

Table 2
Physical and Spectral Data of Compounds 11, 13 and 14

11, 13, 14

Compoun No.	id R	Yield (%)	Mp (°C) (recrystallization solvent)	IR (cm ⁻¹) ${}^{1}H$ NMR (δ ppm) (DMSO-d ₆) [a]	,	MS m/z (R.I. %)	Molecular Formula	Analysis(%) Calcd./Found		
							C	Н	N	
11	Н	45	>300	3070, 1710,	3.28 (s, 3H, 3-CH ₃),	M+ 297	C ₁₄ H ₁₁ N ₅ O ₃	56.57	3.73	23.56
			(DMF)	1680, 1620,	3.49 (s, 3H, 1-CH ₃),	(100)	., ., .	56.55	3.80	23.68
				1510, 1300,	7.30-8.20 (m, 4H, Ar-H).					
				1230, 750.						
13	CH ₃	40	>300	1720, 1640,	3.27 (s, 3H, 3-CH ₃),	M+311	$C_{15}H_{13}N_5O_3$	57.88	4.21	22.50
	•		(DMF)	1610, 1580,	3.51 (s, 3H, 1-CH ₃),	(2),		57.95	4.19	22.39
				1500, 1420,	3.96 (s, 3H, 11-CH ₃),	15				
				740.	7.40-8.30 (m, 4H, Ar-H).	(100)				
14	CH ₂ SCH ₃	42	>300	1720, 1680,	2.25 (s, 3H, S-CH ₃),	M+357	$C_{16}H_{15}N_5O_3S$	53.77	4.23	19.60
			(DMSO)	1650, 1600,	3.24 (s, 3H, 3-CH ₃),	(3),		53.87	4.30	19.55
				1500, 1220,	3.45 (s, 3H, 1-CH ₃),	61				
				740.	5.78 (s, 2H, CH ₂ -S),	(100)				
					7.40-8.30 (m, 4H, Ar-H).					

10 and anthranilic acid ethyl ester in DMF in the presence of anhydrous potassium carbonate and a small amount of cuprous bromide (Scheme 2). The structure of 11 was confirmed by analytical, ir, ¹H nmr and mass data (Table 2).

When this reaction was performed with an excess of 10 in refluxing nitrobenzene, the N-(2-ethoxycarbonylphenyl)-di-8-theophyllinamine 12 was obtained in 41% yield (Scheme 2). Analytical data, ir and mass (M⁺, m/z = 521) spectra are in agreement with the proposed structure 12.

Compound 12 can be cyclized to 11 in 77% yield by treatment with PPA at 180° (Scheme 2).

Compound 11 was methylated to give 13 in the same conditions employed for the preparation of 8 (Scheme 2, Table 2).

Compound 14 was obtained by refluxing 11 with DMSO as described above for 9 (Scheme 2, Table 2).

When 8-aminocaffeine 15 and o-chlorobenzoic acid were submitted to the Ullmann reaction, only the amine 16 was isolated in 40% yield (Scheme 3). By heating compound 16 with an excess of PPA at 120°, the cyclization product 17, an isomer of 8 and 13, was obtained (Scheme 3).

8-Aminocaffeine 15 was synthesized from 8-aminotheophylline 5 using a method described in literature for the preparation of 7-alkyl-8-amino-1,3-dimethylxanthines [13].

When 8-aminocaffeine 15 and o-chlorobenzoic acid were directly heated with PPA at 120°, the amide 18 was obtained in 44% yield (Scheme 3). Numerous attempts to cyclize 18 to 19 were unsuccessful (Scheme 3), probably because the cyclization reaction at the N(9) of the xanthine nucleus was impeded by the presence of the methyl group at position 3 due to its steric hindrance, thus further confirming the structure proposed for compound 6.

EXPERIMENTAL

Melting points were determined using a Reichert Köfler hotstage apparatus and are uncorrected. Infrared spectra were obtained on a PYE/UNICAM mod. PU 9561 spectrophotometer in Nujol mulls. Nuclear magnetic resonance spectra were recorded on a Varian CFT-20 spectrometer, unless otherwise reported, using tetramethylsilane (TMS) as the internal standard. Mass spectra were obtained on a Hewlett-Packard 5988 A spectrometer using a direct injection probe and an electron beam energy of 70 eV. Magnesium sulfate was always used as the drying agent. Evaporations were made *in vacuo* (rotating evaporator). Analytical tlc was carried out on Merck 0.2 mm precoated silica gel aluminium sheets (60 F-254). Elemental analyses were performed by our Analytical Laboratory and agreed with theoretical values to within ± 0.4%.

8,10-Dimethylpurino [7,8-a] quinazoline-5,9,11(6H,8H,10H)-trione 6.

A) From 8-Aminotheophylline 5.

A suspension of 8-aminotheophylline 5 (1.95 g, 10 mmoles), 2-chlorobenzoic acid (1.56 g, 10 mmoles), anhydrous potassium

carbonate (1.66 g, 12 mmoles) and 0.05 g of cuprous bromide in 5 ml of DMF was heated at 180° for 16 hours. After cooling, the reaction mixture was diluted with water and acidified with concentrated hydrochloric acid. The crude product was collected and purified by sublimation (270-300°, 0.4 mm Hg). The recrystallization solvent, yield, melting point, analytical, and spectral data are given in Table 1.

B) From 8-(2-Chlorobenzoylamino)theophylline 7.

A suspension of 7 (0.20 g, 0.6 mmole) and anhydrous potassium carbonate (0.25 g, 1.8 mmoles) in 5 ml of DMF was refluxed for 4 hours, and then, after cooling, it was diluted with water and acidified with concentrated hydrochloric acid. The crude product was collected and purified by sublimation (0.091 g, yield 51%).

8-(2-Chlorobenzoylamino)theophylline 7.

8-Aminotheophylline 5 (0.51 g, 2.6 mmoles) and 2-chlorobenzoic acid (0.41 g, 2.6 mmoles) were added to 6.0 g of PPA, and heated at 120° under stirring for 1 hour. The reaction mixture, after cooling, was treated with iced water. The solid obtained was collected, washed with saturated aqueous sodium hydrogen carbonate solution and with water, and then purified by recrystallization from ethanol to give 0.39 g of pure 7 (yield 45%), mp 284-286°; ir: 3170, 1680, 1640, 1500, 1320 cm⁻¹; ¹H nmr (DMSO-d₆): δ 3.27 (s, 3H, 1-CH₃), 3.43 (s, 3H, 3-CH₃), 7.67 (s, 4H, Ar-H), 12.30 (br s, 1H, 7 NH) ppm; ms: m/z (relative intensity) 333 (M+7), 139 (100).

Anal. Calcd. for $C_{14}H_{12}ClN_5O_3$: C, 50.39; H, 3.62; N, 20.98. Found: C, 50.46; H, 3.57; N, 21.10.

6,8,10-Trimethylpurino[7,8-a]quinazoline-5,9,11(6H,8H,10H)-trione 8, and 1,3,11-Trimethylpurino[8,7-b]quinazoline-2,4,6(1H,3H,11H)-trione 13.

A solution of dimethyl sulphate (0.38 ml, 4 mmoles) in acetone (2 ml) was added dropwise, at 0°, to a suspension of anhydrous potassium carbonate (0.138 g, 1 mmole) and of 6 or 11 (0.297 g, 1 mmole) in acetonitrile (5 ml). The reaction mixture was allowed to stir at room temperature for 8 hours. The solid was collected, washed with water and purified by recrystallization. The recrystallization solvents, yields, melting points, analytical, and spectral data are given in Tables 1 and 2.

8,10-Dimethyl-6-methylthiomethylpurino[7,8-a]quinazoline-5,9,11(6H,8H,10H)-trione 9, and 1,3-Dimethyl-11-methylthiomethylpurino[8,7-b]quinazoline-2,4,6(1H,3H,11H)-trione 14.

A solution of 6 or 11 (0.199 g, 0.67 mmole) in DMSO (5 ml) was heated at reflux for 10 hours. After cooling, the solid collected proved to be a mixture of the starting material and of the methylthiomethyl derivative 9 or 14, respectively. Separations of compounds 9 and 14 from the corresponding starting materials were carried out by preparative tlc on Merck 2 mm precoated silica gel glass plates (60-F 254) using chloroform:methanol = 9.5:0.5 as the eluting system. The solid recovered was purified by recrystallization (Tables 1 and 2).

1,3-Dimethylpurino[8,7-b]quinazoline-2,4,6(1H,3H,11H)-trione 11.

A) From 8-Bromotheophylline 10.

A suspension of 8-bromotheophylline 10 (0.78 g, 3 mmoles), ethyl 2-aminobenzoate (0.61 g, 3.7 mmoles), anhydrous potassium carbonate (0.51 g, 3.7 mmoles) and a small amount of cuprous bromide in 5 ml of DMF was heated at 180° for 70

hours. After cooling, the suspension was treated with water and acidified with concentrated hydrochloric acid. The solid was collected and purified by sublimation (270-300°, 0.4 mm Hg). The recrystallization solvent, yield, melting point, analytical, and spectral data are given in Table 2.

B) From N-(2-Ethoxycarbonylphenyl)-di-8-theophyllinamine 12.

A mixture of 12 (1.5 g, 2.9 mmoles) and 15 g of PPA was heated at 180°, under stirring, for 40 minutes. The reaction mixture, after cooling, was treated with iced water and the solid formed was collected. A further amount of the product could be obtained by neutralization with concentrated ammonia of the acid mother liquor. The solids collected were combined and purified by sublimation and recrystallization from DMF obtaining 0.656 g (yield 77%) of pure 11.

N-(2-Ethoxycarbonylphenyl)-di-8-theophyllinamine 12.

A suspension of 8-bromotheophylline 10 (2.95 g, 10 mmoles), ethyl 2-aminobenzoate (0.5 ml, 10 mmoles), anhydrous potassium carbonate (1.49 g, 10 mmoles) and a small amount of cuprous bromide in 3 ml of nitrobenzene was heated at 220° for 15 hours. After cooling, the reaction mixture was treated with water and acidified with concentrated hydrochloric acid. The solid was collected, washed with petroleum ether 40-60° and ethanol, and purified by recrystallization from DMF to give 2.15 g (yield 41%) of pure 12, mp >300°; ir: 1690, 1660, 1610, 1560, 1520, 1290, 1210 cm⁻¹; ms: m/z (relative intensity) 521 (M+12), 29 (100).

Anal. Calcd. for $C_{23}H_{23}N_9O_6$: C, 52.97; H, 4.45; N, 24.17. Found: C, 53.03; H, 4.60; N, 24.08.

8-Aminocaffeine 15.

Sodium hydride (0.984 g, 41 mmoles, dispersion in mineral oil) was added in small portions to a hot suspension (90°) of 7.02 g (36 mmoles) of 8-aminotheophylline 5 in 300 ml of DMF. After cooling at room temperature, iodomethane (2.25 ml, 36 mmoles) was added dropwise, and the reaction mixture was left under stirring for 16 hours. The solid obtained was collected and washed with petroleum ether 40-60° and water to give 6.48 g (yield 86%) of 8-aminocaffeine 15, mp >300°; ir: 3370, 3300, 3200, 1670, 1620, 1520, 1210, 1020, 950 cm⁻¹; 1 H nmr (DMSO-d₆): δ 3.16 (s, 3H, 1-CH₃), 3.34 (s, 3H, 3-CH₃), 3.54 (s, 3H, 7-CH₃), 6.84 (br s, 2H, NH₂) ppm; ms: m/z (relative intensity) 209 (M+74), 82 (100).

Anal. Calcd. for $C_8H_{11}N_5O_2$: C, 45.93; H, 5.30; N, 33.48. Found: C, 46.01; H, 5.35; N, 33.33.

N-(2-Carboxyphenyl)-8-caffeinamine 16.

A suspension of 8-aminocaffeine 15 (0.418 g, 2 mmoles), 2-chlorobenzoic acid (0.313 g, 2 mmoles), anhydrous potassium carbonate (0.276 g, 2 mmoles) and copper powder (0.07 g) in 3 ml of DMF was refluxed for 16 hours. After cooling, the reaction mixture was treated with water and acidified with concentrated hydrochloric acid. The solid was collected, and purified by sublimation (270-300°, 0.4 mm Hg) and recrystallization from DMF to give 0.263 g (yield 40%) of pure 16, mp >300°; ir: 1680, 1640, 1600, 1540, 1240, 1210 cm⁻¹; 1 H nmr (recorded on a Bruker AC-200) (DMSO-d₆): δ 3.24 (s, 3H, 1-CH₃), 3.47 (s, 3H, 3-CH₃), 3.81 (s, 3H, 7-CH₃), 7.10-8.60 (m, 4H, Ar-H) ppm; ms: m/z (relative intensity) 329 (M+ 76), 15 (100).

Anal. Calcd. for C₁₅H₁₅N₅O₄: C, 54.71; H, 4.59; N, 21.27.

Found: C, 54.63; H, 4.43; N, 21.15.

1,3,5-Trimethylpurino[8,9-b]quinazoline-2,4,11(1H,3H,5H)-trione 17.

A mixture of compound 16 (0.33 g, 1 mmole) and 5 g of PPA was heated, under stirring, at 180° for 1 hour. After cooling, the reaction mixture was treated with iced water. The solid obtained was collected, washed with saturated aqueous sodium hydrogen carbonate solution and water, and then purified by recrystallization from DMF to give 0.20 g (yield 66%) of pure 17, mp 292-294°; ir: 1700, 1640, 1620, 1590, 1320 cm⁻¹; ¹H nmr (recorded on a Bruker AC 200) (DMSO-d₆): δ 3.31 (s, 3H, 3-CH₃), 3.75 (s, 3H, 1-CH₃), 3.84 (s, 3H, 5-CH₃), 7.20-8.20 (m, 4H, Ar-H) ppm; ms: m/z (relative intensity) 311 (M+69), 15 (100).

Anal. Calcd. for $C_{15}H_{13}N_5O_3$: C, 57.88; H, 4.21; N, 22.50. Found: C, 57.86; H, 4.26; N, 22.39.

8-(2-Chlorobenzoylamino)caffeine 18.

A mixture of 8-aminocaffeine 15 (0.54 g, 2.6 mmoles) and 6.0 g of PPA was heated at 120°, under stirring, for 1 hour. The reaction mixture, after cooling, was treated with ice water. The solid obtained was collected, washed with saturated aqueous sodium hydrogen carbonate solution and water, and then purified by recrystallization from ethanol to give 0.40 g (yield 44%) of pure 18, mp 228-230°; ir: 3150, 1700, 1670, 1640, 1530, 1280 cm⁻¹; ¹H nmr (DMSO-d₆): δ 3.27 (s, 3H, 1-CH₃), 3.43 (s, 3H, 3-CH₃), 3.83 (s, 3H, 7-CH₃), 7.60 (s, 4H, Ar-H), 11.40 (s, 1H, NH) ppm; ms: m/z (relative intensity) 347 (M⁺ 6), 139 (100).

Anal. Calcd. for C₁₅H₁₄ClN₅O₃: C, 51.81; H, 4.06; N, 20.14. Found: C, 51.98; H, 4.11; N, 20.05.

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